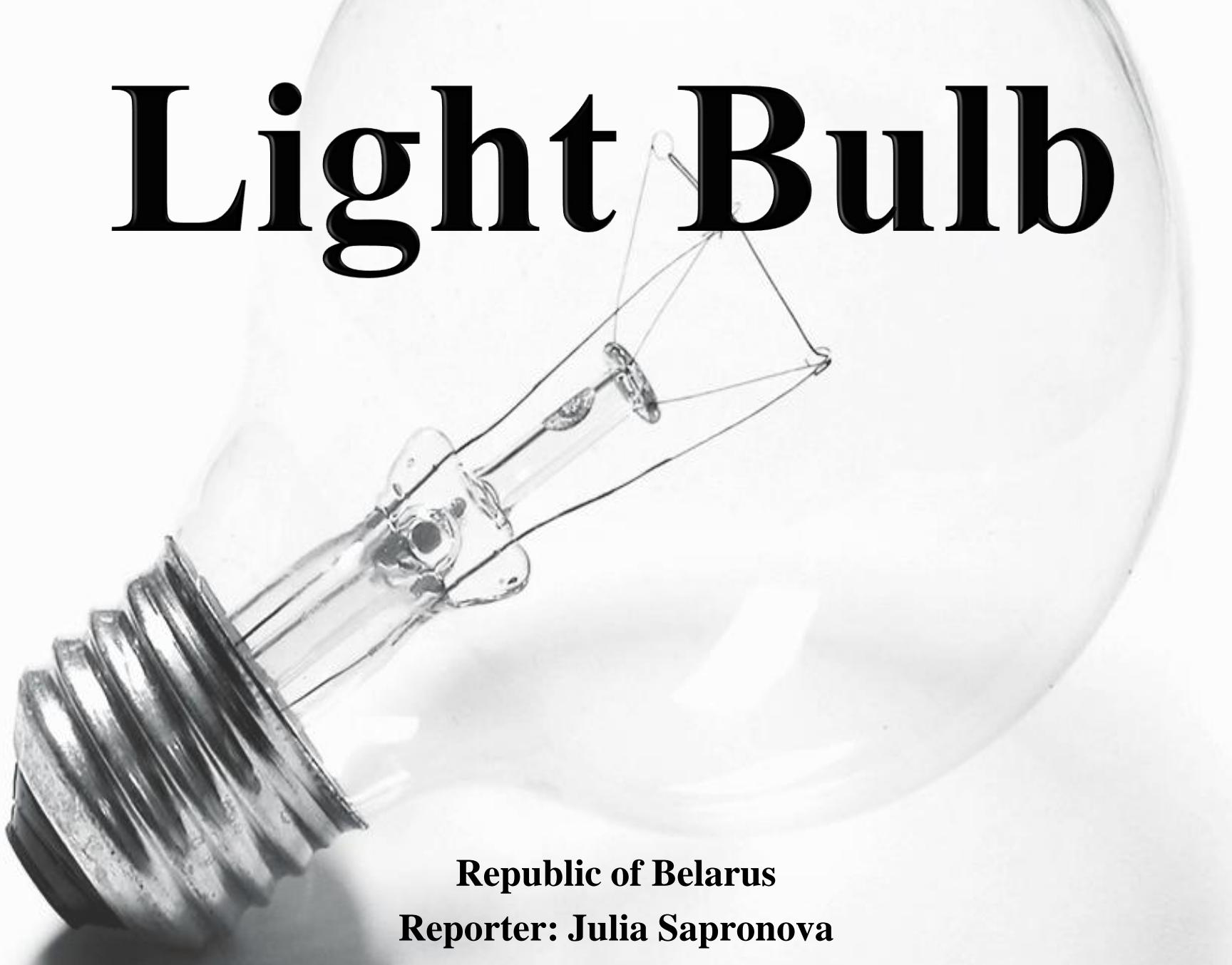


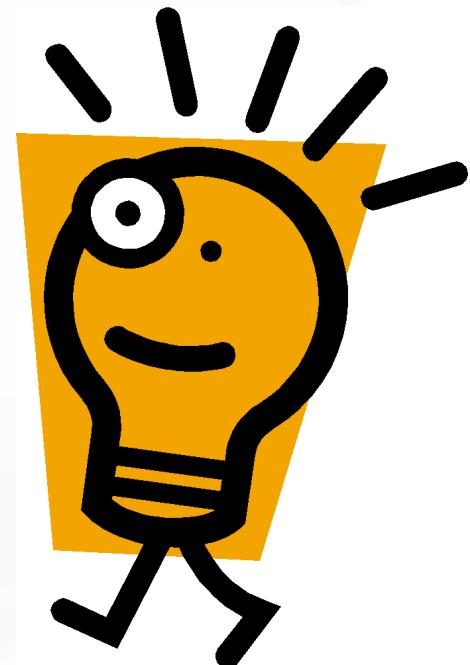
# Light Bulb



**Republic of Belarus  
Reporter: Julia Sapronova  
2011**

# Light Bulb

What is the ratio between  
the thermal energy and  
light energy emitted from  
a small electric bulb  
depending on the voltage  
applied to a bulb?



# Plan of work



Explanation of energy transfer and emission



Theory



Optical part



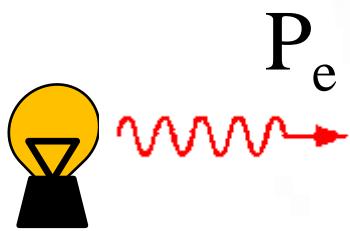
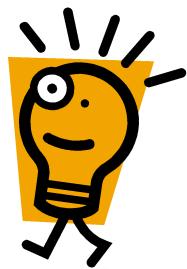
Electrical circuit



Experiment



Conclusions



$$P_o = UI$$

U - voltage

The ratio between the thermal energy and light energy emitted from light bulb

EMITTING **WHAT IS THE HEAT TRANSFER**

**ENERGY**

Light emitting

Infrared emitting

emitting

$$\gamma = \frac{P_o}{P_e}$$

Efficiency of a bulb

Heat

conductivity  
(in wires)

$$\eta = \frac{P_L}{P_o}$$



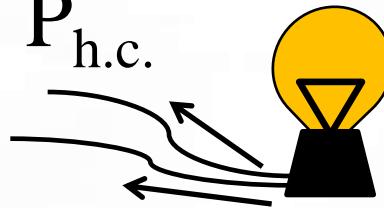
$\lambda - (380\text{nm};$   
 $780\text{nm})$

$\lambda - (380\text{nm};$   
 $200\mu\text{m})$

$P_L$

$P_{inf}$

$P_{h.c.}$



# Equilibrium state



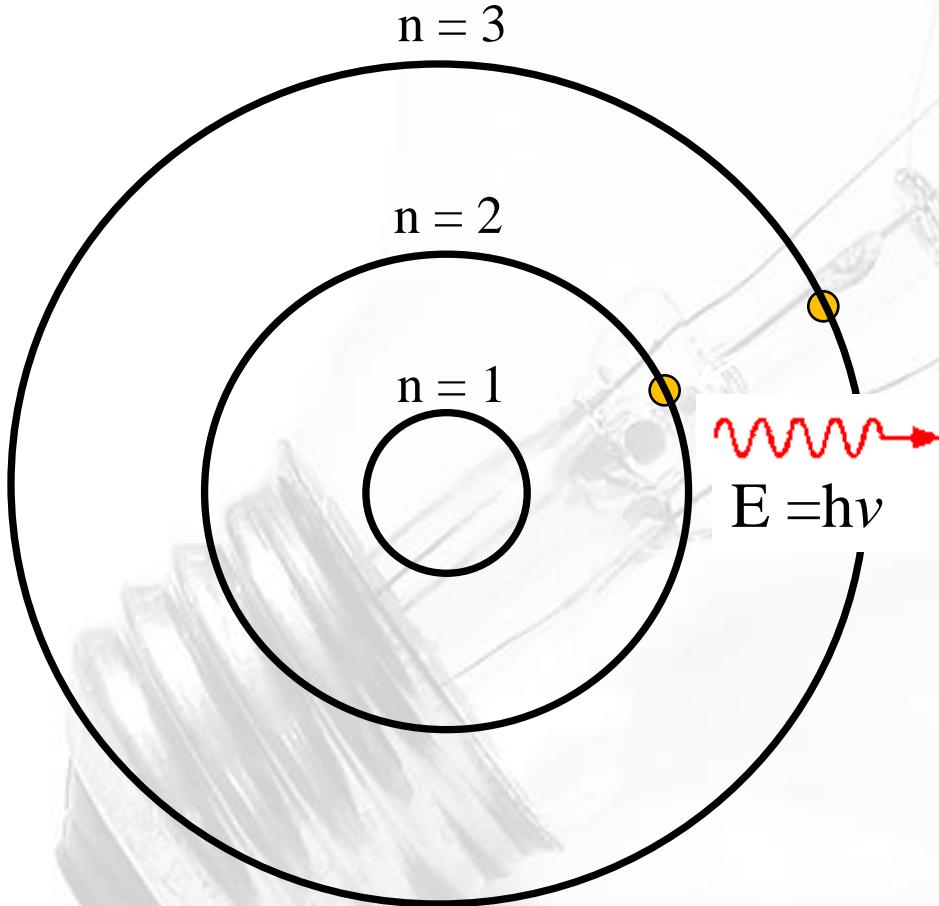
Explanation

Black body

Experiment

Conclusions

# Formation of the emission



Atom gets extra energy

Atom becomes excited

Electron crosses from the highest to the lowest energy level and emits amount of energy - photon

# Planck's law

Planck's law shows dependence of emission intensity on frequency of wave and temperature of emissive body

$$\varepsilon(v, T) = \frac{2hv^3}{c^2} \frac{1}{e^{\frac{hv}{kT}} - 1}$$

**ε** is emissive intensity of blackbody near the light source;

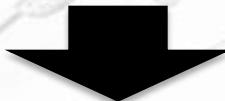
**v** is frequency; **T** is temperature of emissive body;

**h** is Planck's constant; **k** is Boltzmann constant;

**c** is speed of light;

# Planck's law for wavelength

$$\varepsilon(v)dv = \varepsilon(\lambda)d\lambda$$



$$\varepsilon(\lambda) = \frac{2hc^2}{\lambda^5} \frac{1}{e^{\frac{hc}{\lambda kT}} - 1}$$

# Determination the emissive power using Planck's law

Emissive power in range of wavelength from to  $\lambda_i$  to  $(\lambda_i + d\lambda)$

$$P = \varepsilon(\lambda) d\lambda$$

$$P_{vis} = \int_{380nm}^{780nm} \varepsilon(\lambda) d\lambda \quad \text{Emissive power of visible emission}$$

$$P_{inf} = \int_{780nm}^{2700nm} \varepsilon(\lambda) d\lambda \quad \text{Emissive power of infrared emission}$$

# Blackbody

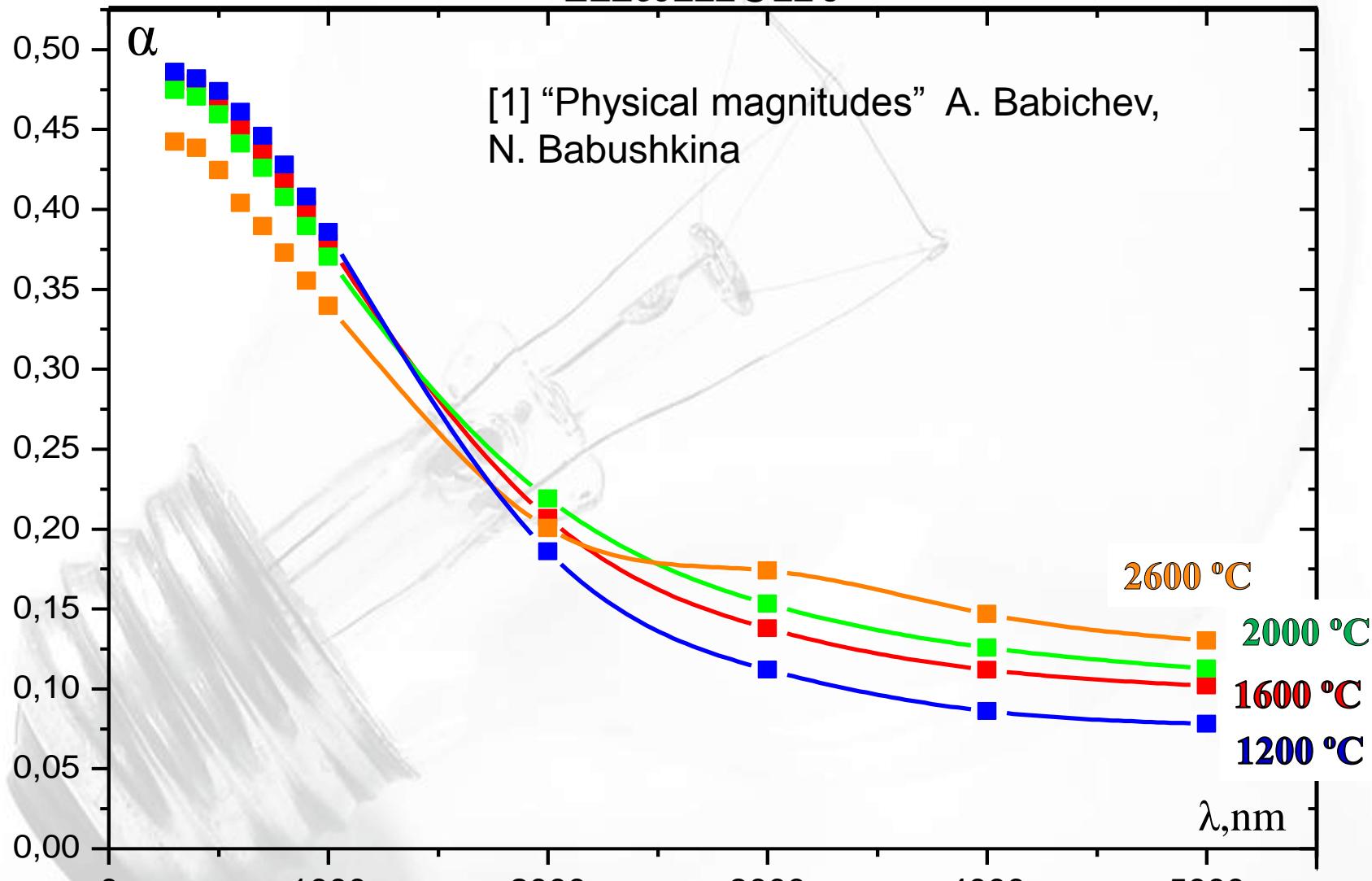
A **blackbody** is an idealized physical body that absorbs all incident electromagnetic radiation



Real body

$\alpha$  is emissivity

# $\alpha(\lambda)$ for different temperatures of tungsten filament



Explanation

Tungsten

Experiment

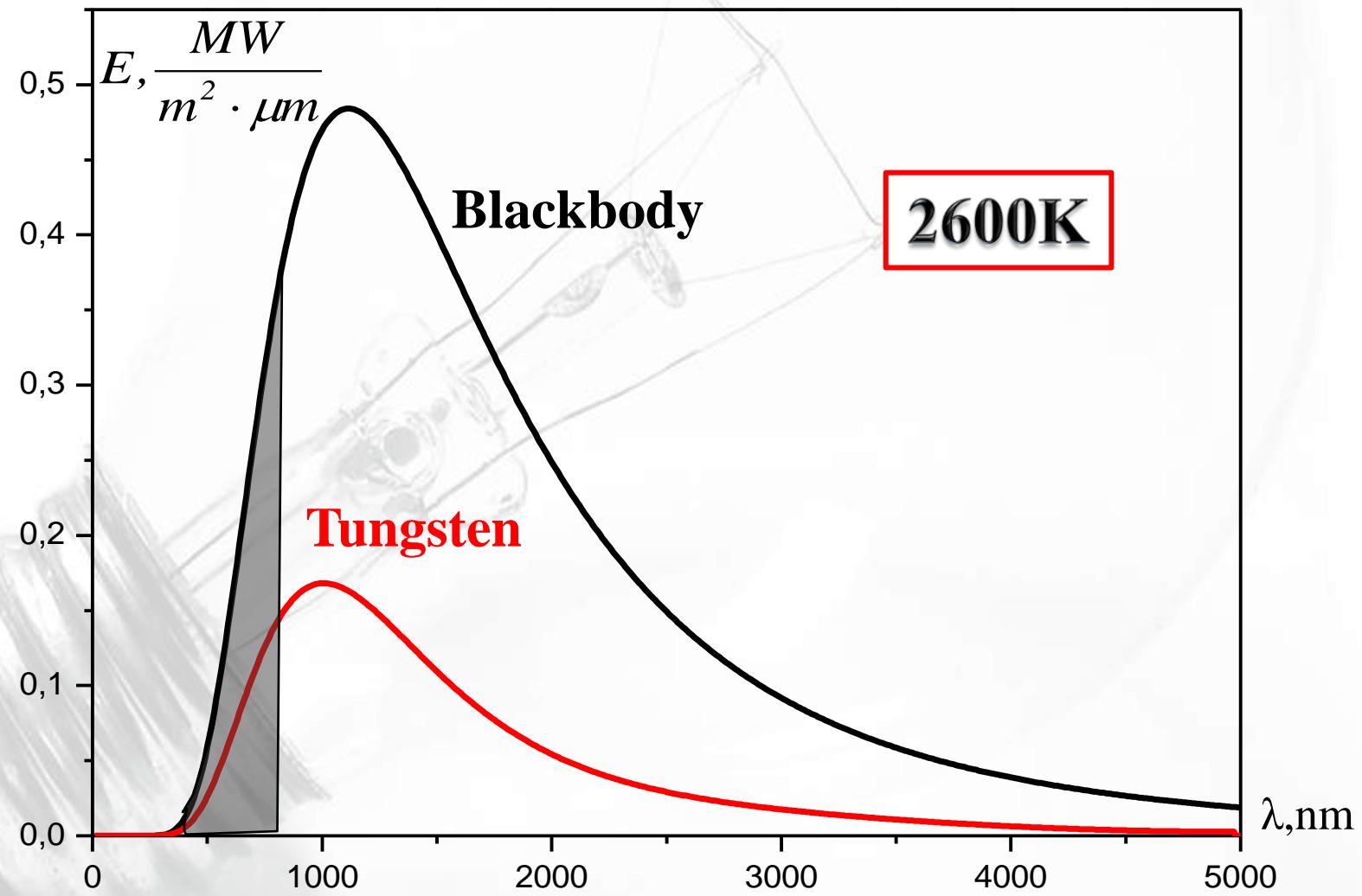
Conclusions

# Dependence $E(\lambda)$ for not blackbody

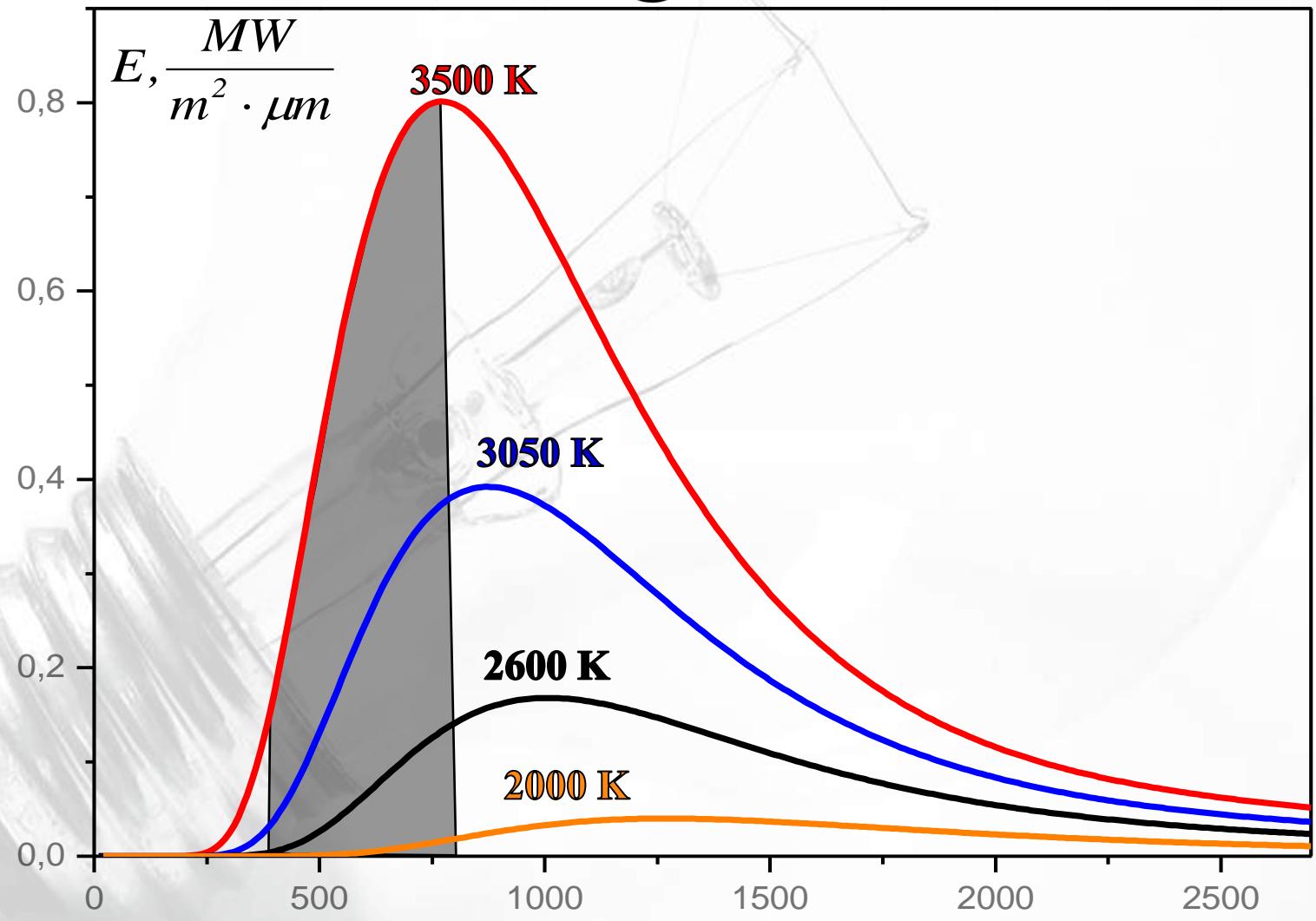
$$E(\lambda) = \alpha(\lambda, T) \varepsilon(\lambda, T) = \alpha(\lambda, T) \frac{2hc^2}{\lambda^5} \frac{1}{e^{\frac{hc}{\lambda kT}} - 1}$$

**$\alpha(\lambda, T)$**  is emissivity of tungsten  
which depends of the wavelength of  
emission and temperature of  
emissive body

# Comparison of the emission intensity of tungsten and blackbody



# Emission intensity on the wavelength for tungsten



# Light bulb without glass



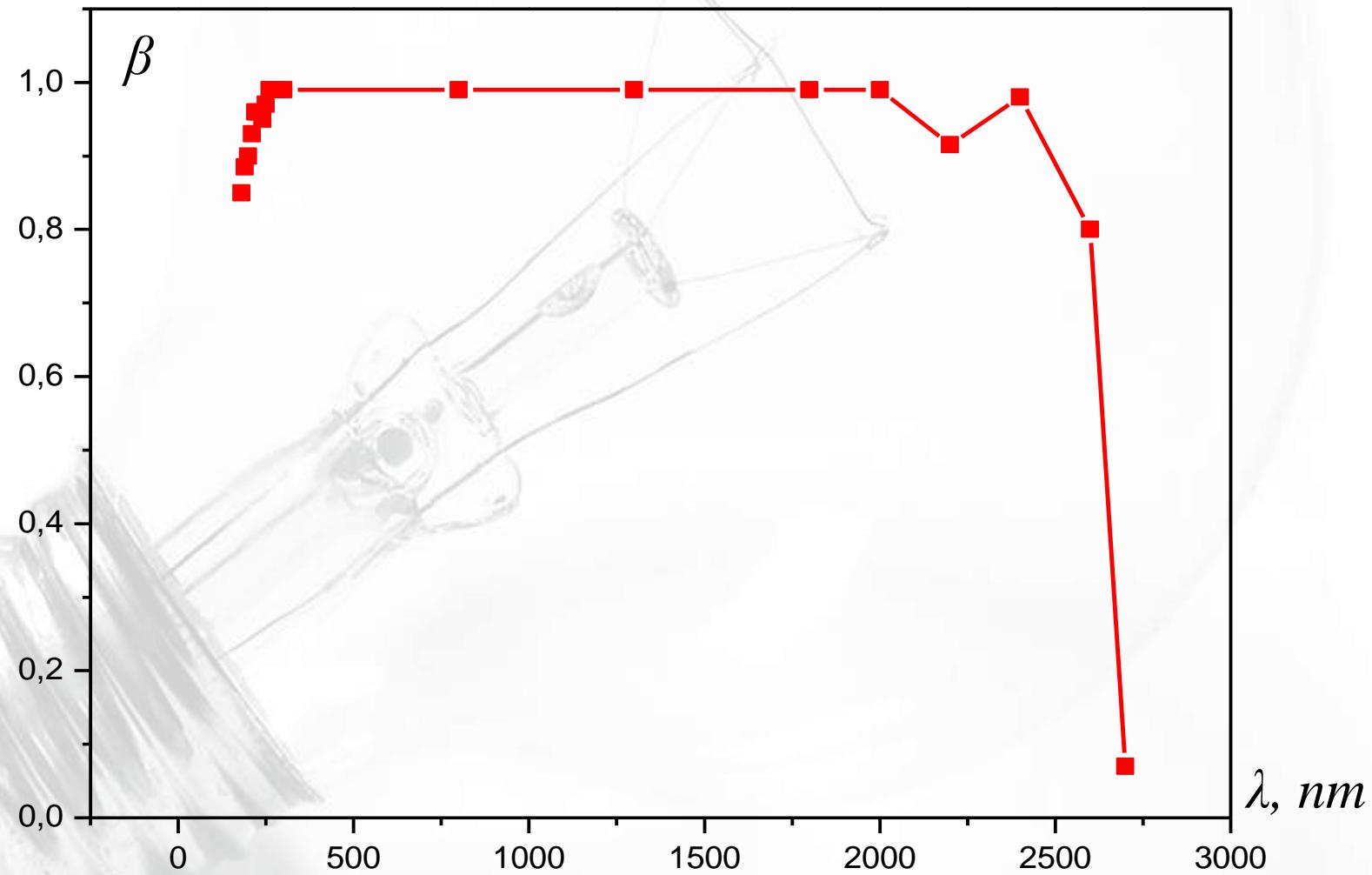
Explanation

Light bulb

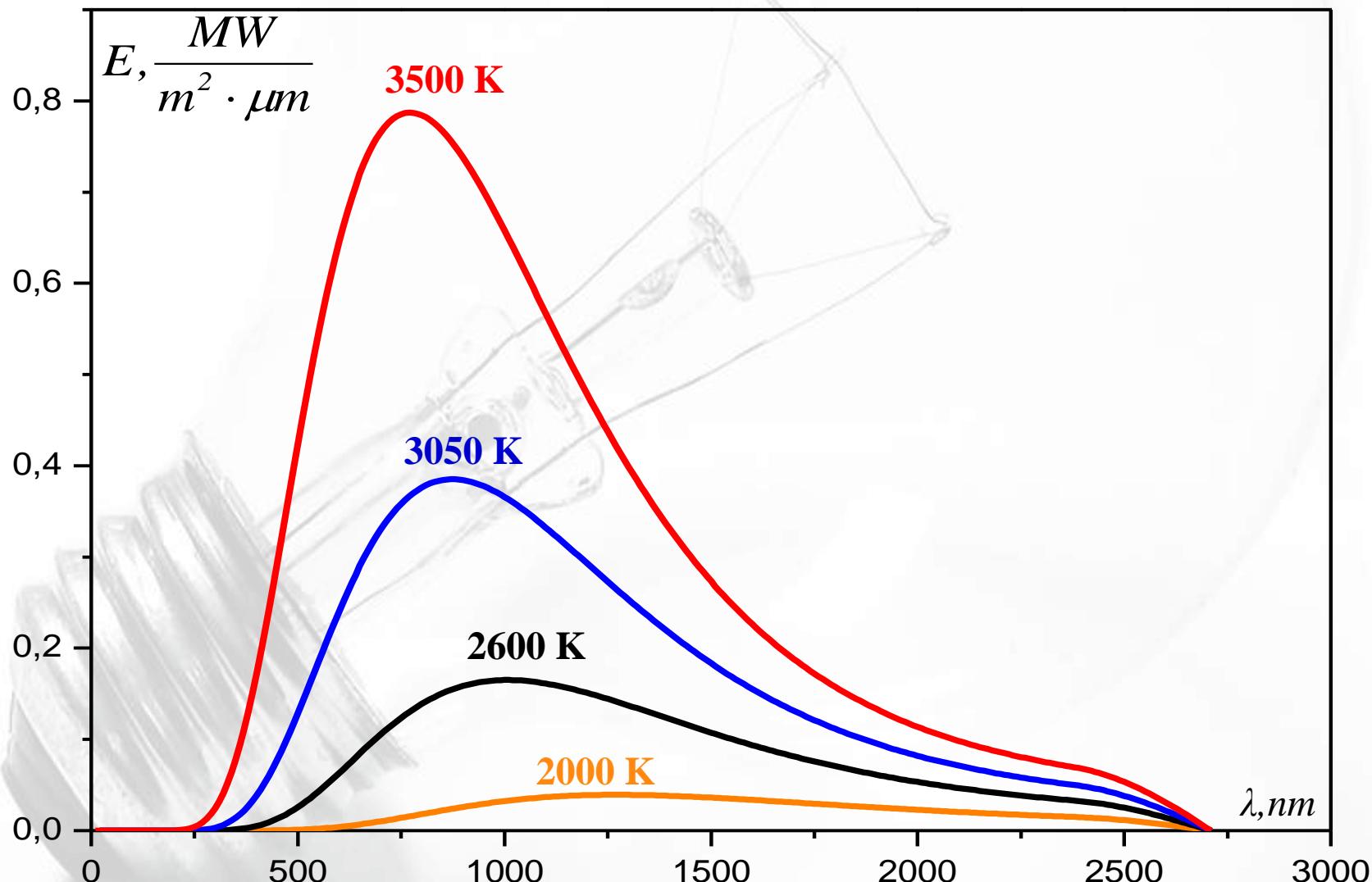
Experiment

Conclusions

# Transmissivity of the glass for different wavelength



# Dependence of emission intensity on the wavelength for light bulb



Explanation

Light bulb

Experiment

Conclusions

# Calculation of $\gamma$

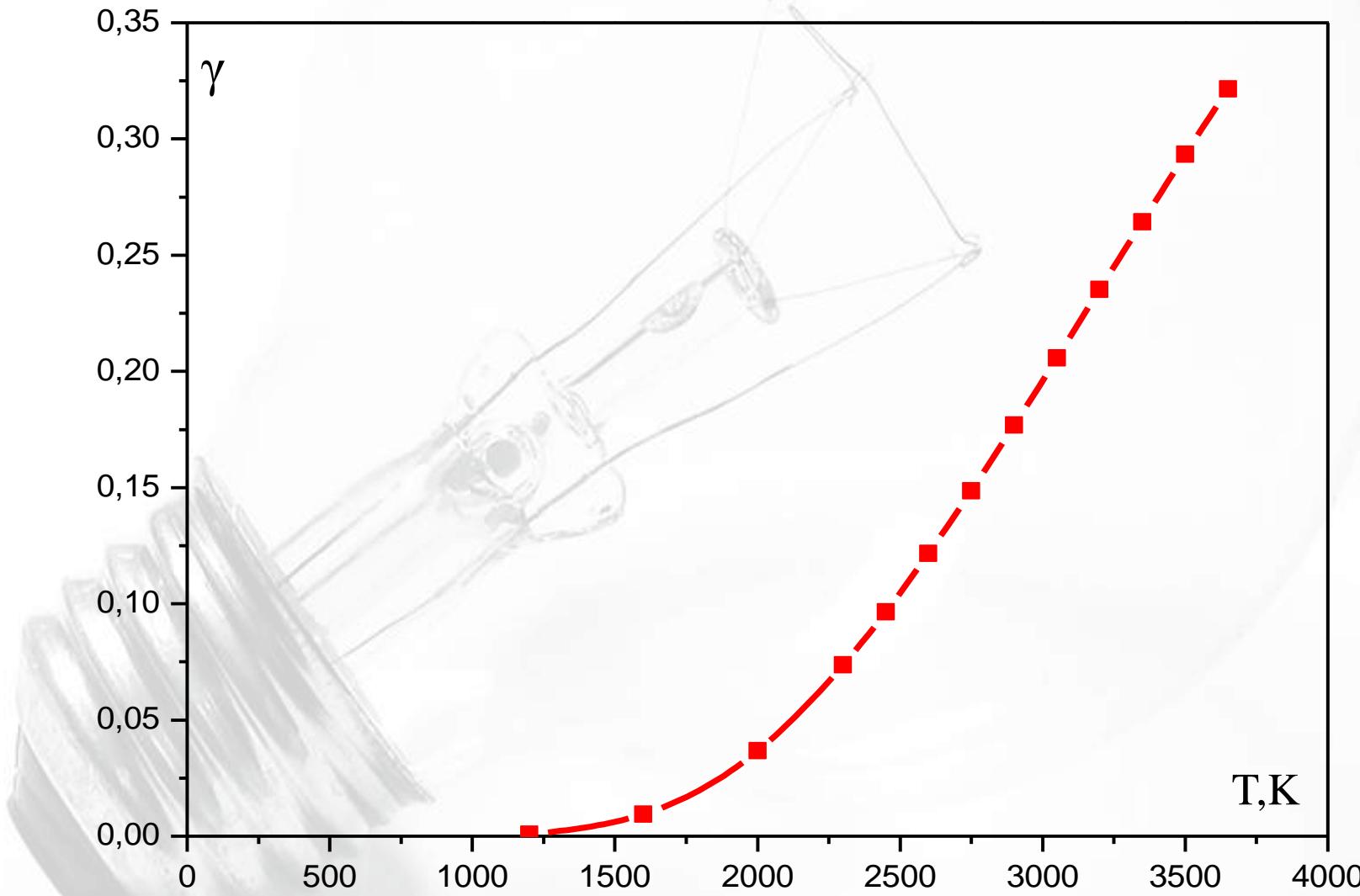
$$\gamma = \frac{\int_{380nm}^{780nm} \alpha(\lambda, T) \beta(\lambda) \varepsilon(\lambda) d\lambda}{\int_{380nm}^{2700nm} \alpha(\lambda, T) \beta(\lambda) \varepsilon(\lambda) d\lambda}$$

$\varepsilon$  is emission intensity of blackbody;

$\alpha$  is emissivity of tungsten;

$\beta$  is glass transmissivity;

# Theoretical dependence of $\gamma$ on the temperature of the tungsten filament



# Making the dependence $\gamma(u)$

$$UI = P_c + P_t + P_e$$

Difficult to estimate precisely

**U** is voltage; **I** is current;

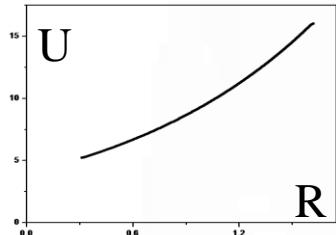
**P<sub>c</sub>** is power of convection heat transfer;

**P<sub>t</sub>** is power of thermal conductivity heat transfer;

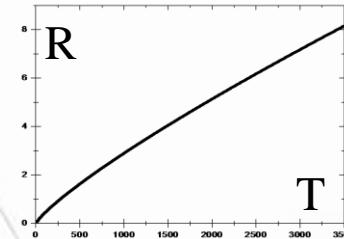
**P<sub>e</sub>** is emissive power

# Making the dependence $\gamma(u)$

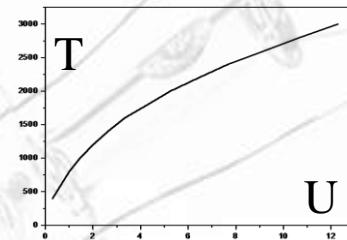
The dependence  $U(R)$



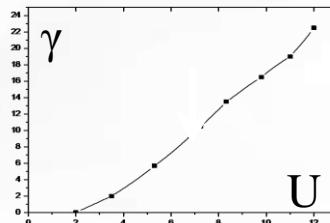
The dependence  $R(T)$



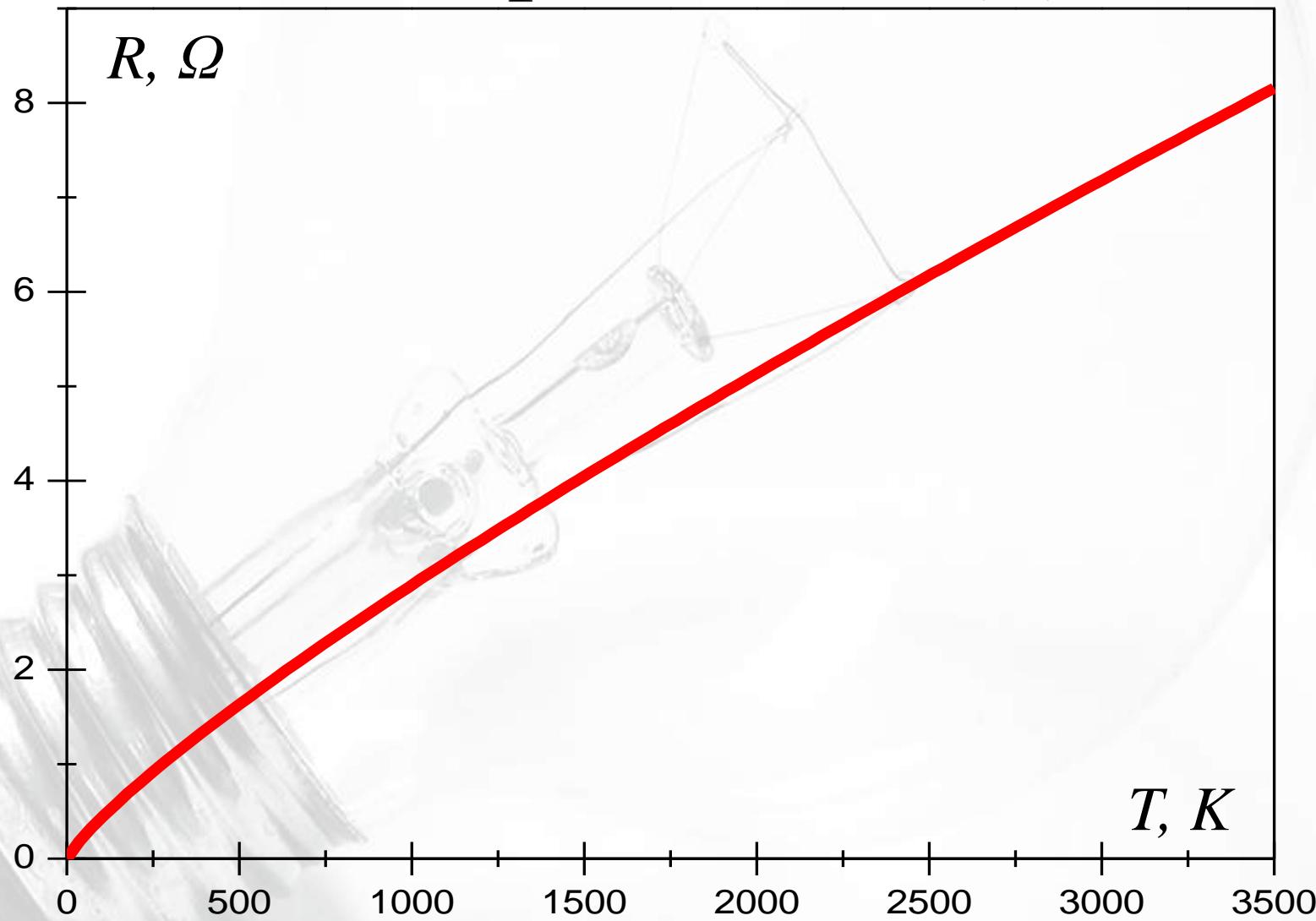
The dependence  $T(U)$



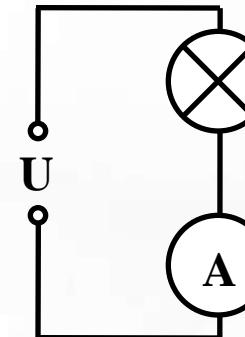
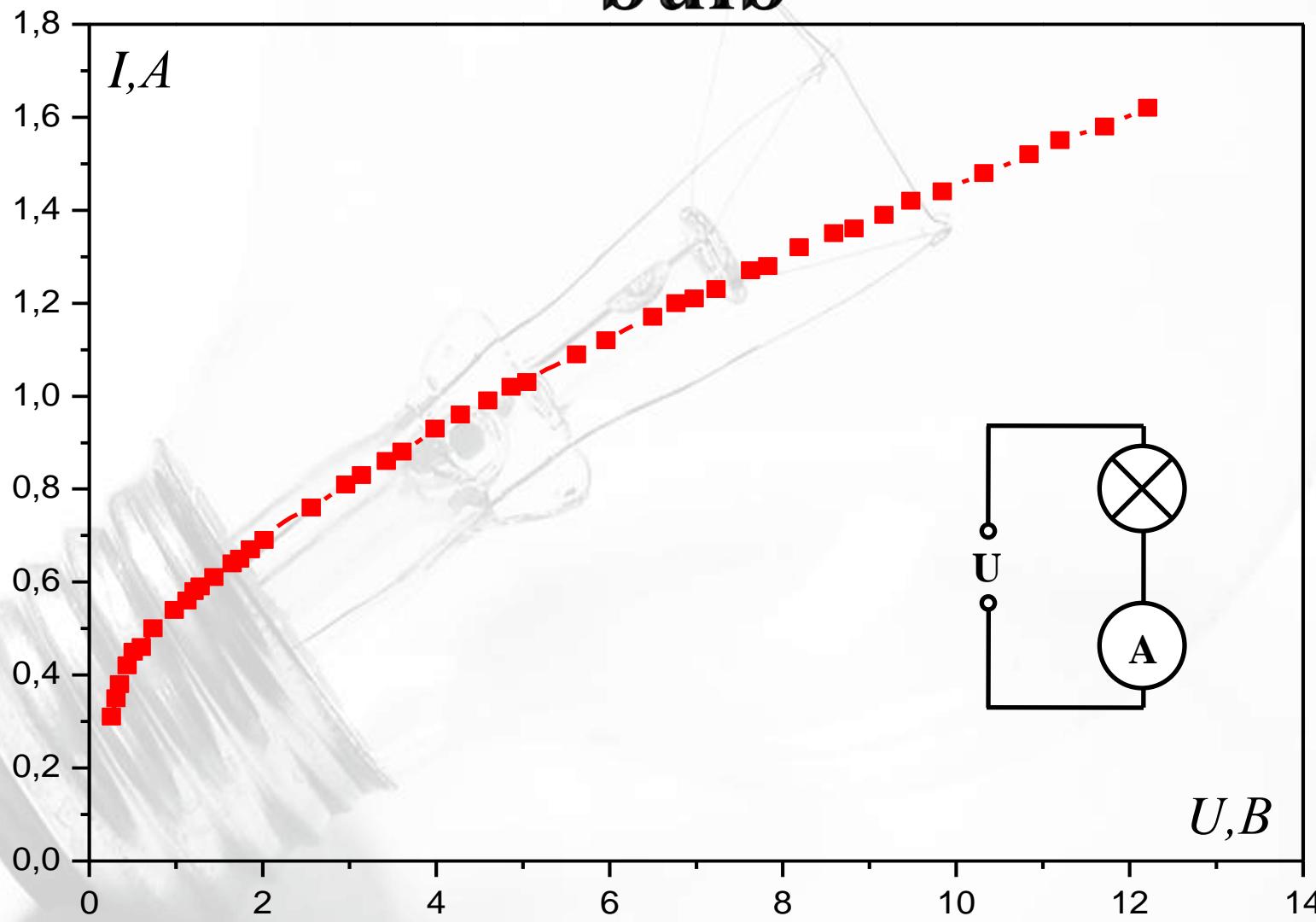
The dependence  $\gamma(U)$



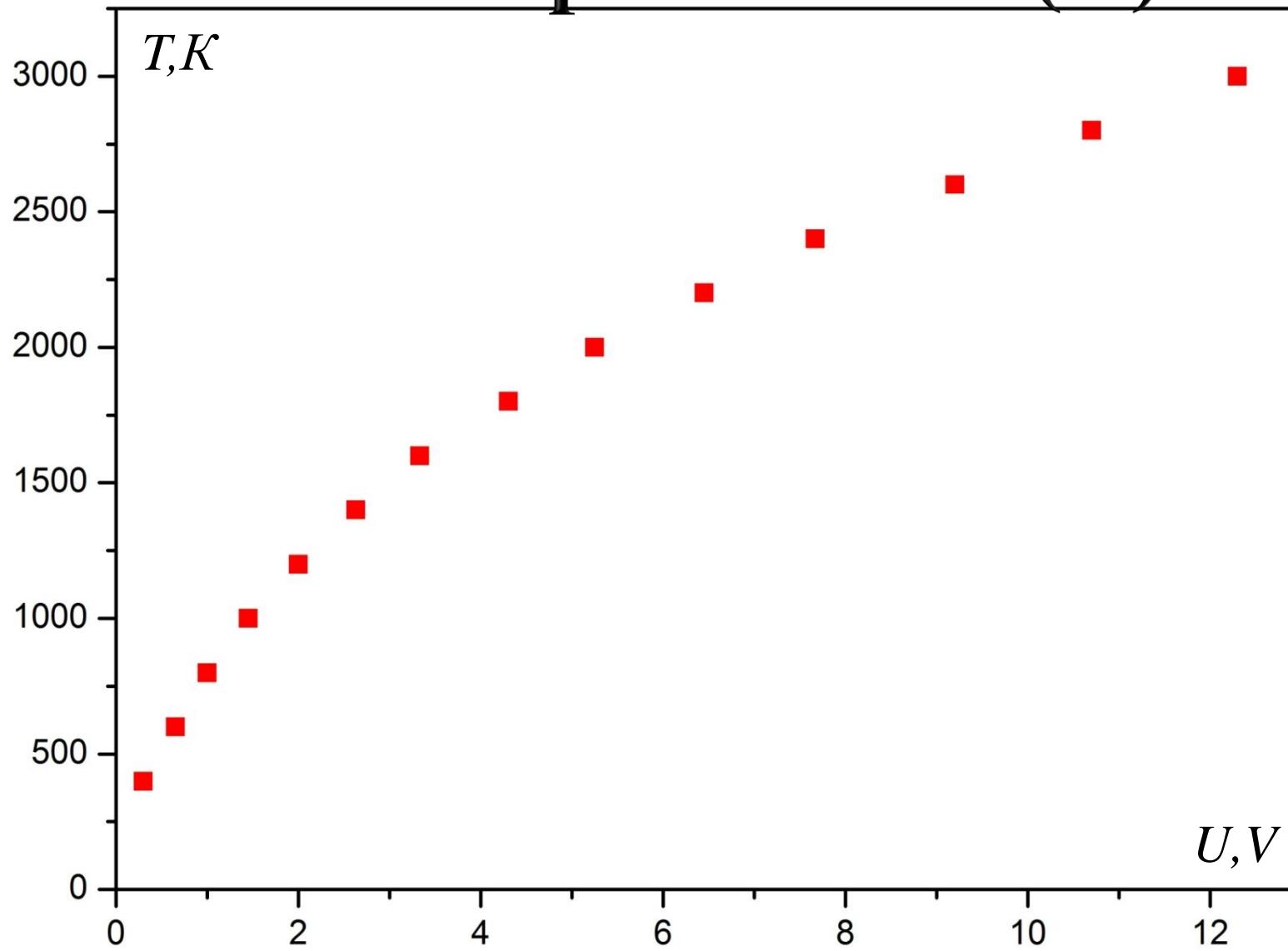
# The dependence $R(T)$



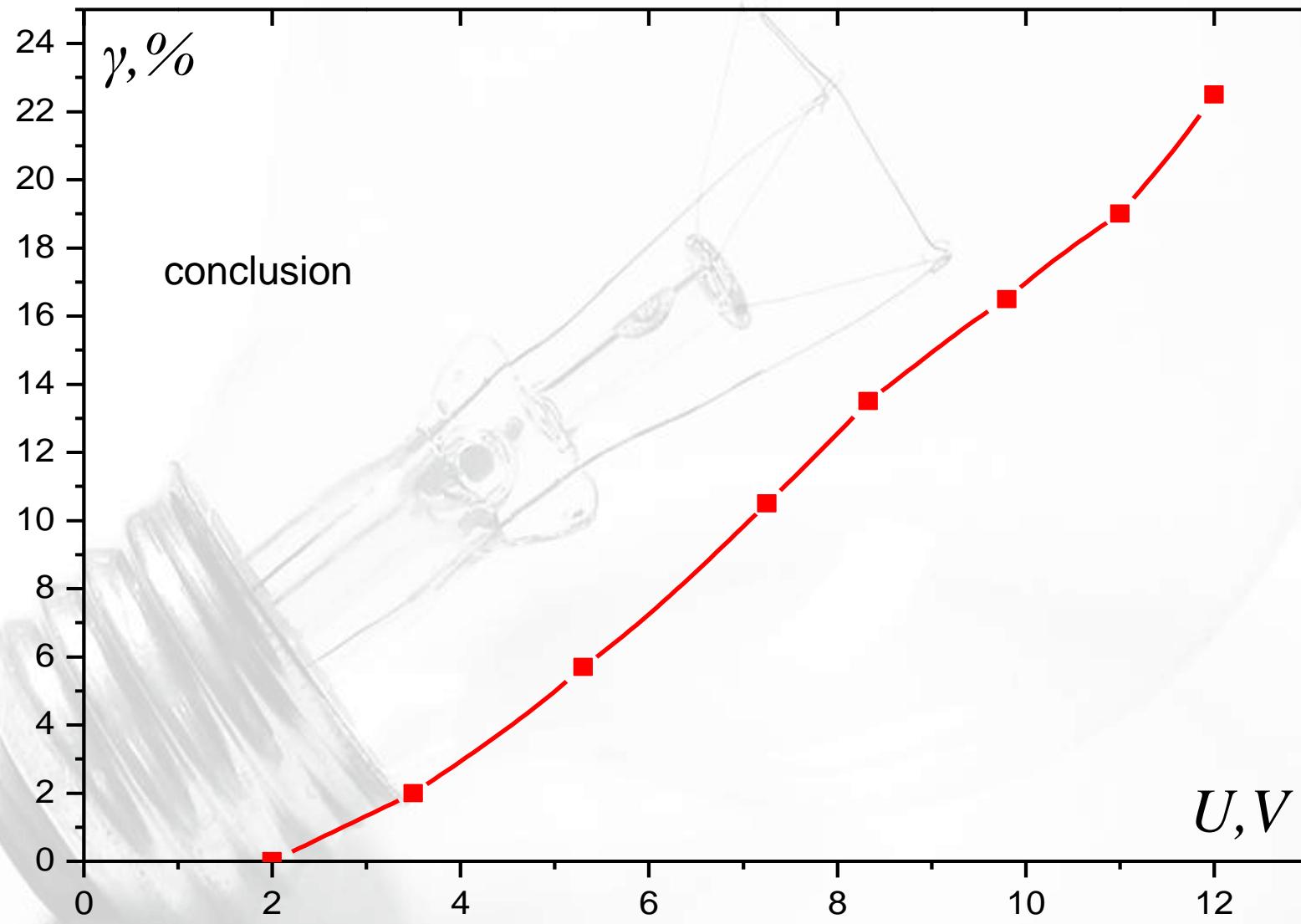
# Volt-ampere characteristic for light bulb



# The dependence T(U)



# THEORETICAL DEPENDENCE $\gamma(U)$



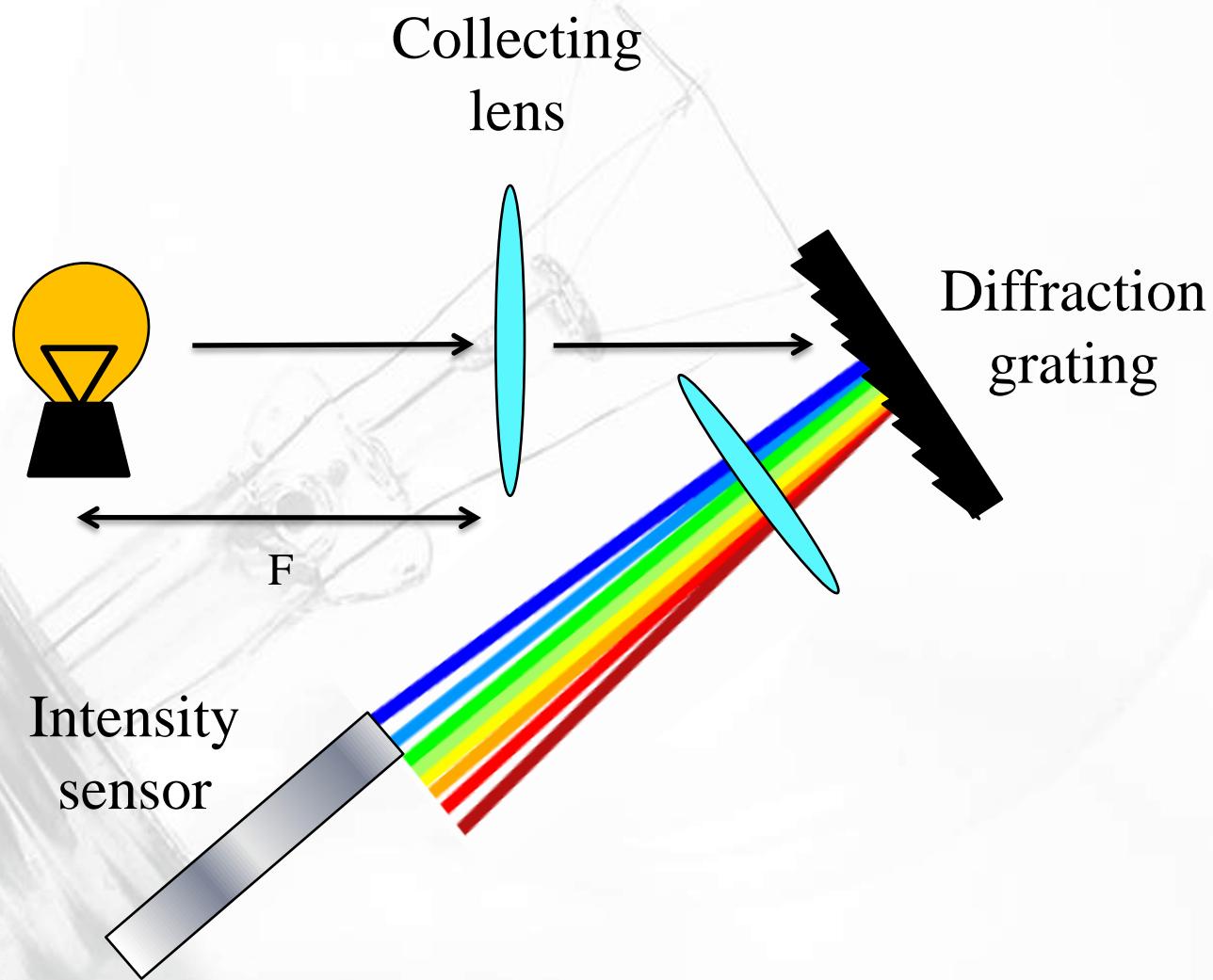
Explanation

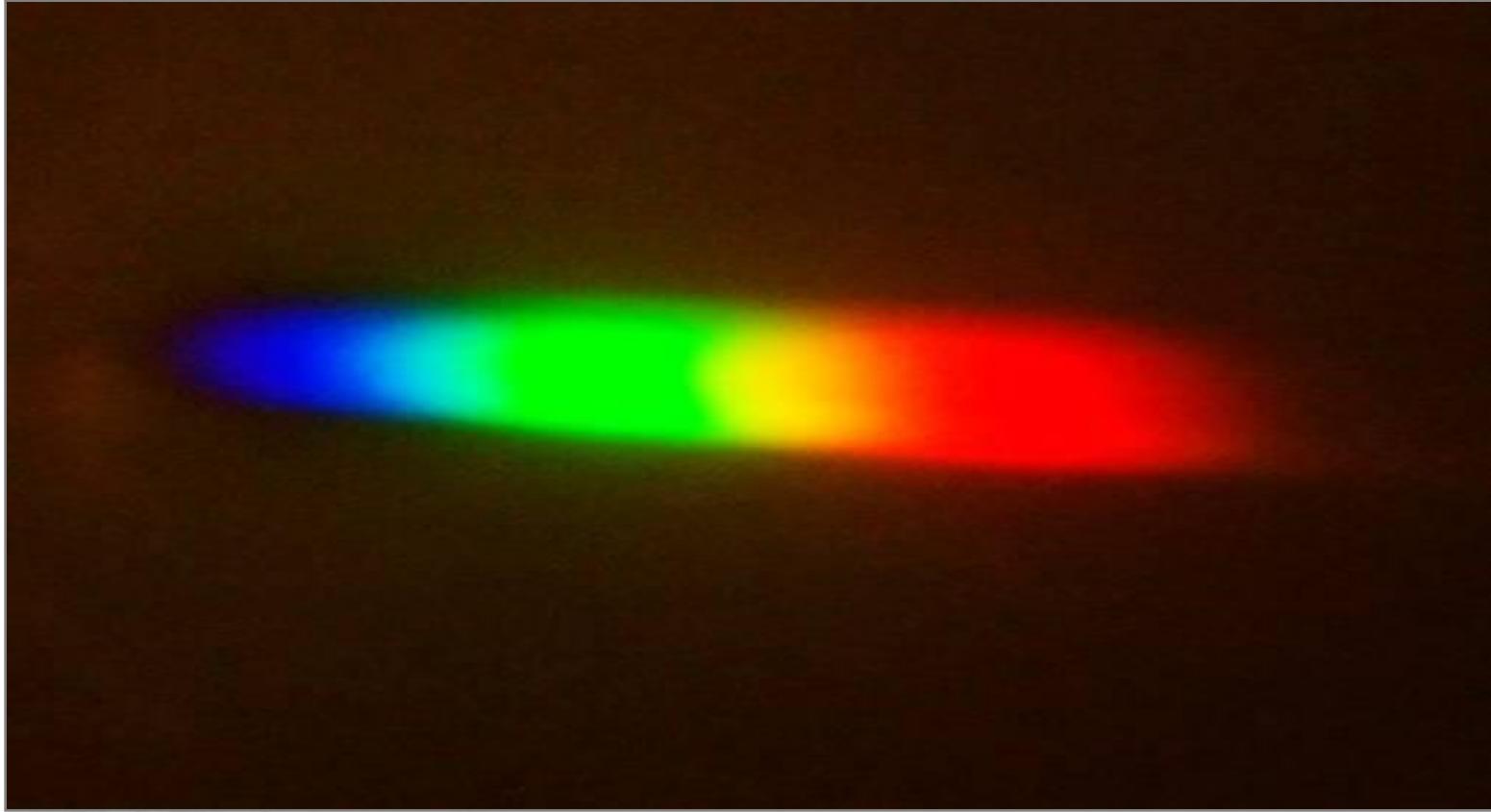
Electrical circuit

Experiment

Conclusions

# EXPERIMENTAL SETUP

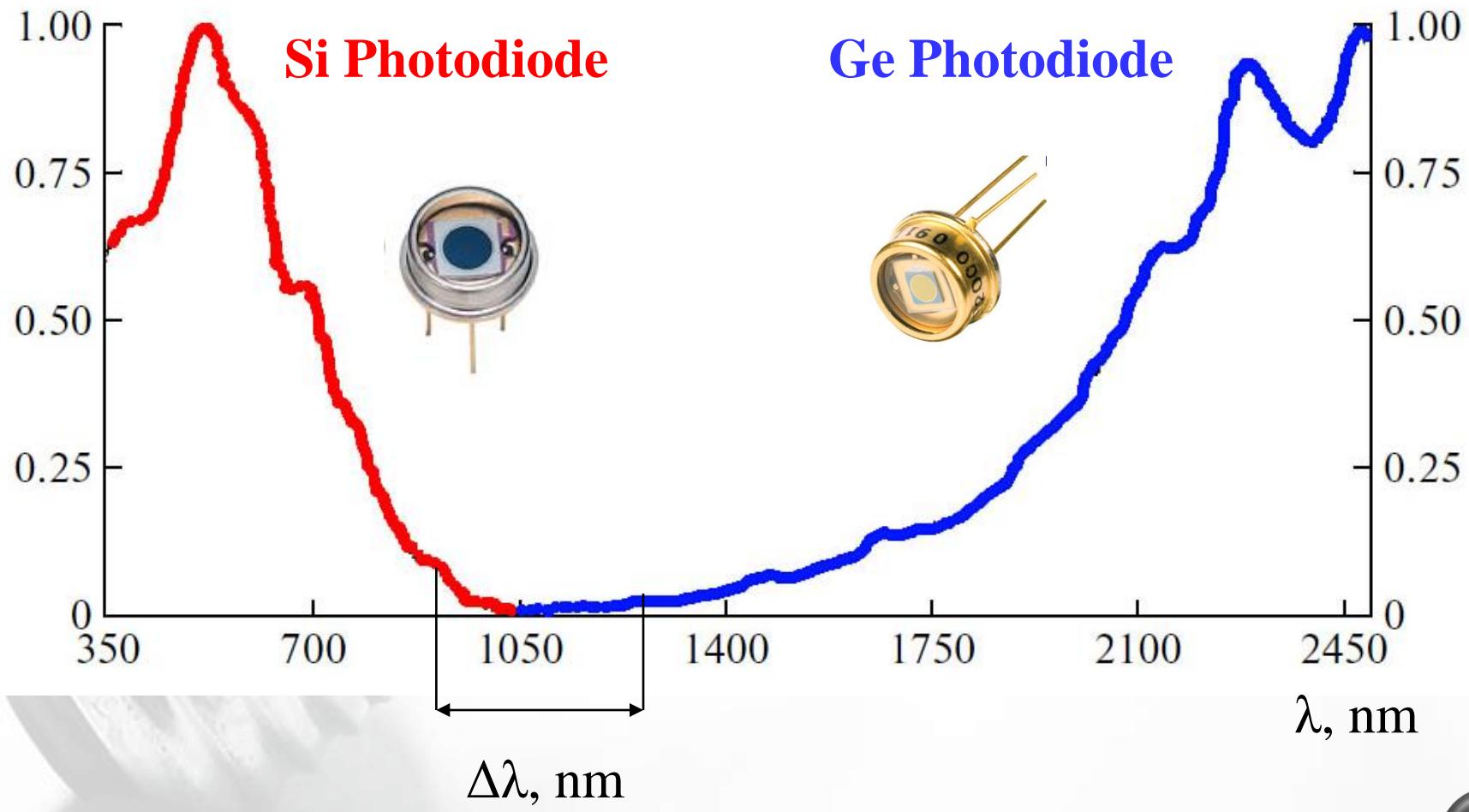




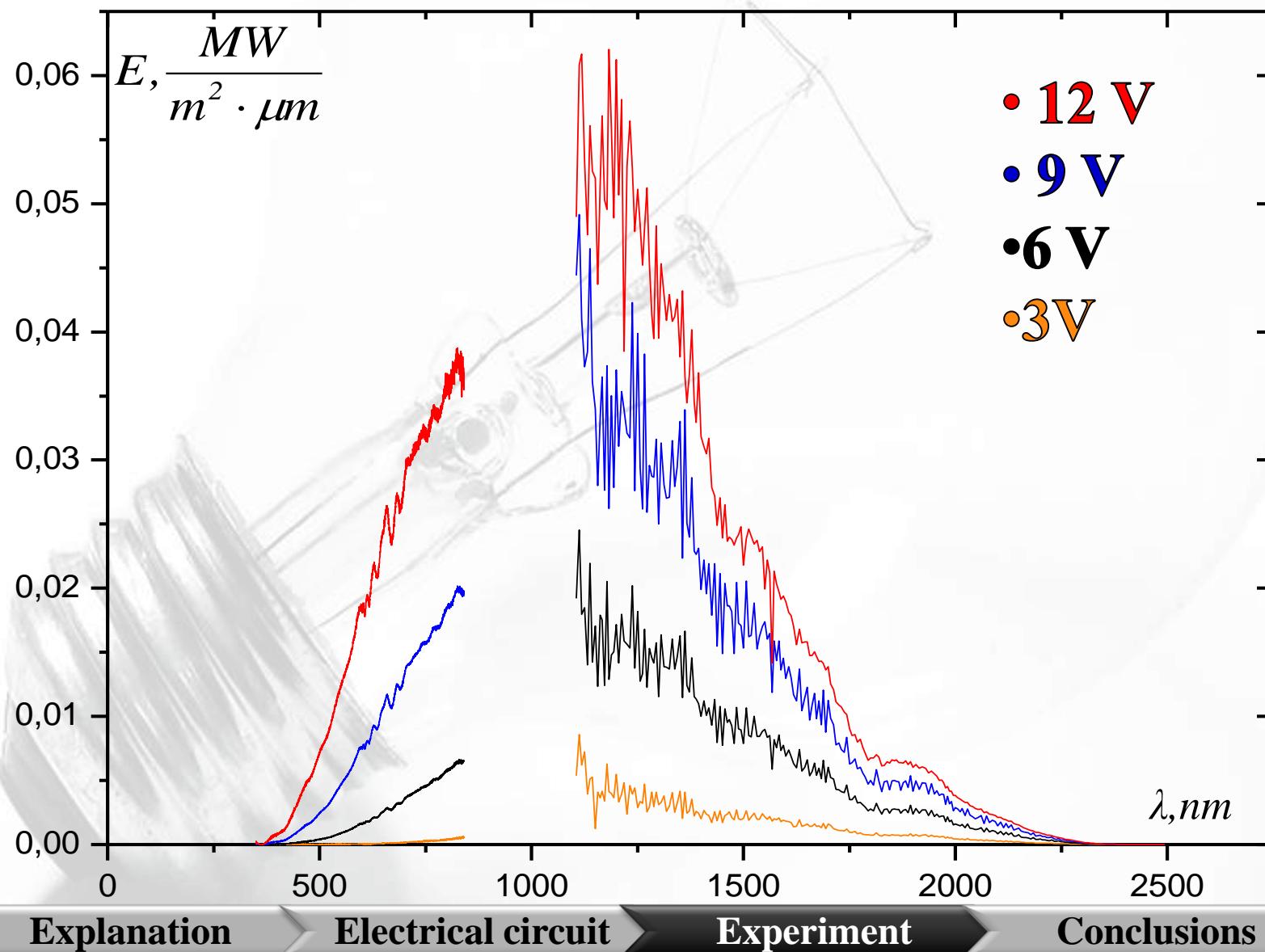
Spectrum obtained from light bulb using collecting lens and diffraction grating

# Intensity measurement

Relative sensibility



# Experimental dependence $E(\lambda)$



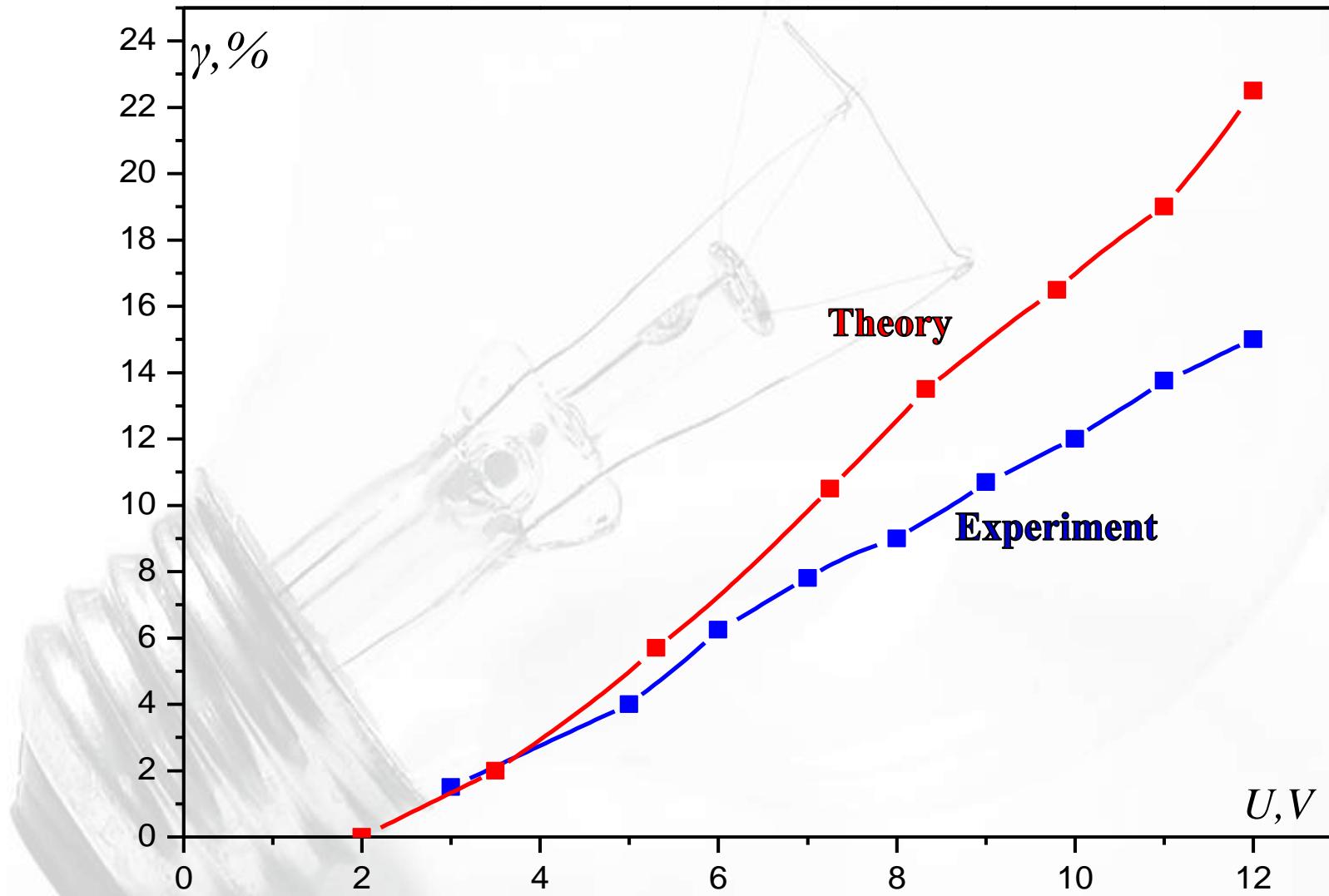
Explanation

Electrical circuit

Experiment

Conclusions

# Comparison of theory and experiment



# CONCLUSIONS

-  The Planck's law can be used for description of emission of a light bulb but emissivity of tungsten and transmissivity of glass should be considered
-  Ratio between thermal and light energy emitted from a light bulb depends only on the temperature of the tungsten filament
-  Voltage must be increased for increasing the ratio between thermal and light energy
-  The rated voltage is the optimal voltage for light bulb
-  Maximal ratio between thermal and light energy for bulb with rated voltage 12V is  $\gamma=15\%$



**THANK YOU FOR YOUR  
ATTENTION**

Team is supported by



# DETERMINATION OF THE TEMPERATURE OF TUNSTEN FILAMENT

$$R = R_0 \frac{T}{T_0}^{1.21}$$

$R_o$  – electrical resistance for the temperature  $T_o$

$$T_o = 293 \text{ K}$$

$$R_o = 1,05 \Omega$$